	Mineral Name	Diagnostic Features			
	CHALCOPYRITE	metallic; H,3.5-4; S,greenish-black; softer than Pyrite!			
. Ks IIC	COPPER	metallic; tarnishes to greenish-white			
Alwa [,] Metal	GALENA	metallic; shiny gray; high density			
∠ ₩ ∑	MAGNETITE	metallic; black; magnetic; Also called "Lodestone"			
	PYRITE	metallic; brass-yellow; S,dark gray; "Fool's Gold"			
88	CORUNDUM	H,9; hexagonal; dark reddish-brown			
<u>8</u>	OLIVINE	green; "sugary" texture			
c	ORTHOCLASE	white, tan, or salmon pink; H,6; Cl. in two directions at right angles; also Potassium Feldspar or "K-spar"			
Thai	PLAGIOCLASE	alternating light and dark color			
U O	QUARTZ	H,7; Hexagonal crystals; or smoky, white or pink			
ard	TOPAZ	white; H,8; Cl. in one direction			
Ţ	TOURMALINE	Black; H,7-7.5, vertical striations on crystal faces			
	APATITE	green or brown; hexagonal			
<u>a</u>	FLUORITE	H,4; Cl. four directions			
7 <u>0</u> 8	HEMATITE	dull or metallic; S,reddish-brown; "bloodstone"			
offe G	LIMONITE	duli or metallic; rusty spots; S,yellow-brown			
ဟိ	MALACHITE	green; very thin striations (layers)			
	BIOTITE	black; splits in thin sheets; a type of Mica			
<u> </u>	CALCITE	reacts with acid; H,3; Cl. in 3 directions			
Ĕ	CINNABAR	scarlet red color; S,scarlet/orangish red			
ny Ge	GYPSUM	can be splintery; H,2			
Softer Penny	HALITE	salty taste			
ys S A	KAOLINITE	chalky texture; H,1			
Alway	MUSCOVITE	transparent; splits in thin sheets; a type of Mica			
Ā	SULFUR	Yellow; smell resembles rotten eggs			
	TALC	soapy or greasy feel; H,1			

Mineral Identification Notes

Α	is a substance	that is naturally	occurring, inorganic, and solid,	with a characteristi	c atomic structure
and v	with a chemical composition and	physical propert	ies that are either fixed or that	vary within specific	limits.
Mine	erals are different from	Rocks ar	e an aggregate or made up of n	ninerals.	
make from	composition of a mineral can be ended a chemical analysis of a mineral the rest are its	to identify it. M	ost common minerals can be id such as hardness or color The first sto	entified by one or m . Properties that se	nore of their t one mineral apart
	ical Properties of Minerals	, ,			
1.	,	ed by evamining	g a fresh surface in reflected lig	ht	
2.		eral's resistance tance of known	e to being scratched. The hardn hardness, or by using it to scrat	ess can be determin ch a substance of k	nown hardness. Ten
	Diamond	10	<u> </u>		
	Corundum	9	The	ere will be a quiz	over this!
	Topaz	8			
	Quartz	7			
	Orthoclase	6	Steel File	(6.5 ±)	
	Apatite	5	Glass	(5.5 ±)	
	Fluorite	4	Nail or Blade	(5.1 ±)	
	Calcite	3	Copper penny	$(3.0 \pm)$	
	Gypsum	2	Fingernail	(2.1 ±)	
	Talc	1			
3. 4.	have either a metallic or nonmeriron, brass, or steel, it has a me Nonmetallic Lusters a. glassy – the lust b. pearly – (talc or c. silky – (gypsum) d. dull – no luster (etallic luster. If a tallic luster. : er of glass (quar muscovite) (kaolinite) he k plate. Streak o	tz) of the mineral. The strea annot be obtained when the m	eshly cut piece of co	ommon metal such as
5.	others have several directions o	f cleavage.	break along planes of weaknes		ve no cleavage while

6.	is when a mineral breaks in a direction other than along a cleavage plane.
	There are several types of fracture:
	Uneven: The surface of the fracture is irregular and rough. Most minerals have this kind of fracture.
	Conchoidal: The surfaces of the fracture are smooth and sometimes rounded.
	Earthy: Soft minerals may exhibit smooth but dull fracture surfaces. Similar to a broken dirt clod. Splintery: A fibrous fracture.
	Spinitery. A librous fracture.
7.	Sometimes the crystal shape is visible and can help set it apart from other
	minerals with similar characteristics.
8.	Minerals, like Calcite, will effervesce (bubble) when treated with dilute
	hydrochloric acid.
_	
9.	. A few minerals are soluble enough to have diagnostic tastes. Halite for example has a strong salty taste.
10	Some minerals have a distinct smell such as Sulfur.
IŲ.	Southe militer als mave a distinct sine is such as summi.
11.	A few minerals will attract a magnet. Example: Magnetite
12.	Minerals like Calcite can split light rays into two parts. One ray travels
	straight through the mineral. The other ray is bent. This causes two images to be seen when viewed through a
	transparent specimen.
13.	The mineral's ability to glow while under ultraviolet light.
De	efine other key terms:
Crs	vstal
Ci }	Juli
_	
Spo	ecific Gravity
Or	e
Ge	m
	····
.	web advan
ı et	rahedron

Structure of Minerals Notes

A)	Minerals are often to	und as	•	
	1. Crystal: A regular _		solid.	
	2. Orderly arrangement			of crystal.
B)	basic cry	stal shapes.		
	1. Described by crysta	llographic	(over)	
C)	Tetrah	edron		
	1. Silicates: compoun	ds of silicon	and oxygen plus one	or more
		_ element.		
	2. Four	_ atoms pack	ked closely around a	silicon atom
	held together by a		bond.	

Crystal Systems

System	Axes of intersection	Ideal	Length of axes
cube or Isometric		W	ali firee axes equal
tetragonal		And district control of the second of the se	two horizontal axes equal third axis different
hexagonal	906		three horizontal axes equal fourth axis different
orthorhombic	.16		all three axes unequal
monoclinic	90° third axis different than 90°	A	all three axes unequal
richnic	all 3 axes different than 90°		all three axes unequal

Name	Date
------	------

3-5 Mineral Word Search

Locate and circle the names of minerals commonly found in earth science textbooks. The names appear in the list below. Names may be found backward, forward, vertically, horizontally, and diagonally. Be careful!

E	Α	C	A	L	С	I	T	E	C	F	P	N	0	A
S	В	H	E	\mathbf{T}	I	\mathbf{T}	A	M	E	H	E	I	L	T
U	G	A	U	0	L	E	C	I	\mathbf{R}	Т	H	Q	E	M
L	\mathbf{T}	L	A	Α	S	В	Ē	L	I	M	A	I	U	G
F	V	C	P	S	E	T	I	R	A	В	W	F	\mathbf{T}	Q
U	C	Ο	I	J	I	K	0	E	A	N	E	L	A	G
R	I	P	В	M	A	U	E	U	W	A	S	C	L	Ė
M	E	Y	0	G	L	A	X	Z	C	G	I	E	С	\mathbf{T}
E	A	R	L	F	R	I	E	Q	N	M	В	P	M	I
A	H	I	Q	M	T	A.	M	C	E	\mathbf{T}	I	N	0	L
C	L	\mathbf{T}	P	E	G	D	L	0	G	C	0	E	L	A
0	F	E	L	D	S	P	A	R	N	Q	\mathbf{T}	C	В	H
\mathbf{R}	X	C	U	M	U	S	P	Y	G	I	I	P	0	\mathbf{T}
Y	Q	U	A	R	T	\boldsymbol{z}	I	0	R	G	T	L	S	I
0	S	1	L	M	E	R	\mathbf{E}	U	P	В	E	E	Т	C
D	T	A	M	0	N	D	Z	A	E	R	U	В	Y	A
S	L	A	0	C	\mathbf{W}	A	E	T	ľ	H	P	A	R	G

AZURITE - Cu3(CO3)2(OH)2

MICA - KAI2(AISi3O10)(OH)2

GYPSUM - CaSO₄2H₂O

FELDSPAR - K(AlSi3O8)

BIOTITE - K(Mg,Fe)1(AlSi1O10)(OH)2

SULFUR - S

LIMONITE - FeO OH HIO

CHALCOPYRITE - CuFeS₂

BARITE - BaSO₄

OPAL - SiO2 · H20

HALITE - NaCl

FLUORITE - CaF:

CALCITE - CaCO₃

TALC - Mg2Si4O10(OH)2

GRAPHITE - C

CHROMITE - FeCr2O4

BAUXITE - AI(OH)

QUARTZ - SiO2

HEMATITE - Fe₂O₃

GALENA - PbS

Use one color for each silicate mineral and another color for each nonsilicate mineral.

Mystery Minerals

A worker at a geology laboratory discovered 5 unlabeled mineral samples in his collection. Two samples were white, one was brown, and two were translucent.

He labeled the samples A, B, C, D, & E. The two white samples, A and B, looked very similar. However, sample A bubbled when dilute hydrochloric acid was poured on it. Sample B did not react to dilute HCl. Sample C, the brown mineral, attracted iron filings and caused the laboratory compass to deflect.

After making these observations, he found 5 labels that had no sample to go with them: Diamond, Quartz, Calcite, Halite, and Magnetite.

Answer the following questions based on the information above:

1.	Based on the lab worker's observations, which is the calcite sample?
	Explain your choice:
2.	Which is the magnetite sample?
	Explain your choice:
3.	What test could the worker perform to clearly identify the diamond sample?
4.	List TWO tests the worker could perform to clearly identify the quartz and halite samples:
	FAMILY TO THE PARTY OF THE PART

Mineral Identification

Match the terms in Column I with the phrases in Column II. Write the letter of the correct phrase in the blank on the left.

Colum	n i	Column II	
	1. cleavage	a. The measure of how easily a mineral can be scratched	
A-4	2. diamon	b. Name given to the scale of hardness	
	3. fracture	c. One of the softest known minerals	
- Lumman	4. hardnes	d. The hardest known mineral	
	5. mica	e. Reflection of light from a mineral's surface	
	6. luster	f. Color left by powdered mineral on unglazed porcelain	
	7. Mohs	g. Tendency to break along smooth, flat surfaces	
	8. quartz	h. A common mineral that breaks along smooth, flat surfaces	
	9. streak	i. Tendency to break with rough or jagged edges	
	10. talc	j. A common mineral that breaks with rough or jagged edges	

In the blank at the left, write the term in the box that correctly completes each statement.

crystal carbonates		halide elements	silicates oxides	4000 98 percent	8 6
	7.	Scientists know	of more than	minerals.	71 - 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1
And the second s	8.	Most of these ar	e composed of onl	y elements.	
	9.	These few eleme	ents make up	of Earth's crust.	
*		Minerals that co		most common rock-form	ming
	11.	A group that inc	ludes rock salt is t	he group.	
	12.	Each mineral ha	s a different	formation.	
	13.	Scientists have i	dentified n	najor formation systems.	
	14.	Two other major	groups of minera	ls are and	

Mineral Identification

Complete the descriptions of the 15 minerals listed below. Using **page 912-913** as reference, match the property in the list with the mineral name.

**All characteristics in the list may be used more than once.

Mineral Characteristics

a. specific gravity of 7 or greater	f. hardness of 6.0	
b. can be a white color	g. source of copper	
c. conchoidal fracture	h. source of iron	
d. hardness of 6.5 or greater	i. contains calcium	
e. hexagonal crystal form	j. contains sulfur	

1.	Copper:	used for wire, pipes, & coins,,
2.	Galena:	cubic cleavage,,
3.	Quartz:	·
4.	Chalcopyrite:	metallic luster,
5.	Limonite:	yellowish brown streak,,
6.	Corundum:	·
7.	Olivine:	green color,,
8.	Plagioclase:	
9.	Pyrite:	metallic luster,
10.	Hornblende:	ferromagnesian silicate,,
11.	Fluorite:	glows under UV light,
12.	Magnetite:	black streak,,
13.	Potassium Feldspa	ar: 2 cleavage planes @ 90°,,
14.	Topaz:	orthorhombic shape,,
15.	Hematite:	·



Minerals Review Questions

						
True	or False.					
	1. The streak test is	a test of mineral density				
	2. Fluorescent mine	rals glow while being sub	jected to ultraviolet	light.		
	3. Cleavage is impor	tant in identifying some	minerals.			
	4. All minerals have	the same basic crystal sh	nape.			
Choo	— ose the best response	·	•			
	ŕ		imate hardness			
	According to the table, what is the approxim of a mineral that scratches quartz and can be			Mineral	Hardness	
	by topaz?	·		corundum topaz	9 8	
	a. 6.5 b.	7.5 c. 8.5	d. 9.5	quartz	7	
				feldspar	6	
	6. A natural, inorgar chemical compo	nic, homogenous solid wi sition is called	th a characteristic		,	
	a. an atom	b. a gemstone	c. a mineral	d. a tetral	nedron	
	_ 7. Minerals that con	tain silicon and oxygen a	re			
	a. sulfides	b. sulfates	c. ores	d. silicates	5	
	8. The appearance o	of light reflected from the	e surface of a mineral	is called		
	a. color	b. streak	c. luster	d. fluores	cence	
	_ 9. The words <i>glassy,</i>	earthy, and dull describe	e a mineral's			
	a. luster	b. hardness	c. streak	d. fluores	cence	
	_ 10. The words <i>uneve</i>	en and splintery describe	a mineral's			
	a. cleavage	b. fracture	c. hardness	d. luster		
	_ 11. Mohs scale is us	ed in measuring a minera	al's			
	a. hardness	b. cleavage	c. color	d. luster		
	_ 12. Double refractio	n is a distinctive property	y of crystals of			
	a. mica	b. feldspar	c. calcite	d. galena		
-	_ 13. A mineral that sp	olits into even sheets sho	ws which of the follo	wing properties?		
	a. low density	b. streak	c. cleavage	d. triclinic	system	
	_ 14. The most comm	. The most common magnetic mineral is				

b. magnetite

c. halite

a. hematite

d. uranium

